



CSL's: A tool
whose time
has come?

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Laboratory Risk Assessment - or - Who Needs a Hood?

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The Laboratory Safety Challenge



- Ever-evolving chemical selection and process
- Lab walls are getting thinner, both figuratively and literally
- Turnover – 20% per year in academia
- The educational culture

Stakeholders in Laboratory Risk Assessment



- Laboratory workers
- Laboratory upper management and planners
- Laboratory designers
- Laboratory building operators
- Emergency responders

Emerging Risks in Lab Buildings



- **Ventilation design and energy costs**
 - Optimum ACH
 - Hood design goals
- **Building operation**
 - Protection of maintenance staff
 - Education of lab staff about their buildings
 - Re- and retro- commissioning
- **Emergency planning and response**
 - Pre-plans
 - Scene assessment
 - Response rate: 1 emergency response/250 lab-years
- **Does this point the way from risk assessment to control banding?**
 - Addressing these issues requires development of broader understanding and language for lab chemical risks

Chemical Risk Assessment



- **Flammability**
 - Concentrations of concern tend to be a few percent by volume
- **Corrosivity**
 - Particles with kinetic energy are much harder to control than vapors
- **Reactivity**
 - Requires chemical and process specific literature review
- **Toxicity**
 - Concentrations of concern range from 1000 ppm to 0.5 ppm to ALARA
- **Implementing GHS will help this process**

Chemical Protection Strategies



- Change the chemical – limited opportunities in the research setting
- Engineering controls
 - General Ventilation
 - Flammable cabinets
 - Local Ventilation
 - Chemical Hoods
- Administrative Controls and Oversight
 - 20% turnover/year in academic labs
- Personal Protective Equipment
 - Of value only in unexpected situations (Wetterhahn and UCLA)
- Traditionally, hood + PPE = “as safe as can be”

Lab Planning and Design: Moving beyond Tradition



- The traditional engineering approach uses a lot of energy without much thought
- Control Banding: Define Chemical Safety Levels from 1 to 4
 - Developed by analogy with Biosafety Levels
- For design and response purposes, the control band deals with a collection of chemicals rather than a specific biological agent
- The result is a general guideline to appropriate protections (which is likely to need modification, e.g CSL 2+)

Factors to Consider in Selecting a CSL:



- **Flammability** (via MSDS):
 - check flashpoint (is it below ambient temperature?)
 - If yes, are expected airborne concentrations above LEL?
- **Corrosivity** (via pH of solutions):
 - pHs < 2 or > 10.5 require special handling
 - Consider also spattering and off-gassing from reactions
- **Reactivity** (via MSDS and literature review)
 - Check potential interactions and contamination concerns
- **Toxicity** (via MSDS and other sources)
 - Review PELs, TLVs, IDLHs and assess against anticipated concentrations
 - Consider potential interactions
 - ALARA for irreversible hazards (cancer, birth defects, sensitization)

Chemical Safety Levels and Lab Facilities



- **CSL-1:** no ventilation
(e.g. cold rooms and warm rooms)
 - Chemical uses similar to residential settings (kitchens and cleaning products)
- **CSL-2:** general ventilation
(X air changes/hour)
 - Chemical uses similar to cars (gallons of flammables and assorted other chemicals)
- **CSL-3:** local ventilation (i.e. hoods)
 - Chemicals similar to hardware stores – emergency concerns is unexpected reactions
- **CSL-4:** high hazard storage or processes that require specialized procedures

Determining the CSL



Hazard	Fire	Corrosivity	Reactivity	Toxicity
CSL 1	Flashpoint below ambient	$2 < \text{pH} < 10.5$	No chemical changes expected in the process	All chemicals have known toxicities and TLV's > 500 ppm
CSL 2	Flashpoint near ambient, expected concentration < 10% LEL	$\text{pH} < 2$ or $\text{pH} > 10.5$	No known incompatibilities between chemicals being used	All chemicals have known toxicities and $10 \text{ ppm} < \text{TLV's} < 500 \text{ ppm}$
CSL 3	Expected concentration > 10% LEL	Use of heated corrosives	Chemicals with known reactions or contamination hazards present	Unknown toxicities or OEL < 10 ppm
CSL 4			High hazard reactions in use	Irreversible toxicities require use of designated areas

But don't forget the other pieces of the puzzle



	Facility	Personal Protective Equipment	Oversight	Emergency Response
CSL 1	Any room, no ventilation	None	Generic self inspection guidelines	Standard response
CSL 2	Ventilated lab room (X ACH)	Nitrile gloves, eye protection	General training and oversight	Fire response
CSL 3	Lab room with local ventilation (fume hood)	Appropriate gloves, eye protection, lab coats	Process specific training and protocols	Hazmat defensive response -> commercial clean-up
CSL 4	Specifically designed lab	Process specific PPE	Written SOPs and specific oversight practices	Specialized hazmat response

Next Steps



- Develop a risk assessment tool that addresses the questions of various stakeholders (web-based?)
- Define the boundaries between the various CSL's and where those boundaries blur
- Develop guidance documents for use of CSL's